

Time: 2:00 Hours

Marks: 50

## Instructions:

- All questions are compulsory.
- Figures to the right indicate marks.

- Que.1 Answer the following question. (Any two) (10)
- Discuss the application of co-ordination compounds in various fields.
  - Explain Werner's theory with limitations.
  - Explain bonding and structure of Zeise salt.
- Que.2 Answer the following question. (Any two) (10)
- Construct multiplication table for  $C_{2v}$  with suitable example.
  - Differentiate  $\sigma_v$ ,  $\sigma_h$  and  $\sigma_d$ .
  - Prove that  $S_n^{2n}=E$  when  $n$  is odd by giving the example of eclipsed ethane.
- Que.3 Answer the following question. (Any two) (10)
- Explain Carnot's cycle with its operation in detail.
  - Derive the equation for entropy change in mixing of ideal gas.
  - Derive Gibbs Helmholtz equation.
- Que.4 Answer the following question. (Any two) (10)
- Explain Dry Process for hydrogenation of oil.
  - Give difference between Enantiomers and Diastereomers.
  - Derive Van't Hoff isotherm equation.
- Que.5 Answer the following question. (Any two) (10)
- (a) Write a note on Ionization isomerism. (03)
  - (b) Give any two uses of organo magnesium compound. (02)
  - (a) Give one difference between  $C_n$  and  $S_n$ . (03)
  - (b) Deduce point group of  $CO_2$  and allene. (02)
  - Two moles of an ideal gas are allowed to expand reversibly and isothermally at 300 K from pressure of 1 atm to 0.1 atm. What is the change in Gibbs free energy ( $R = 1.987 \text{ cal K}^{-1}\text{mol}^{-1}$ )
  - Discuss Saponification value and acid value for oil & fats.

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